

1 Geochemie a radiometrie

wolfram hafnium siderofilní litofilní

rezervoáry: 1 .. mlhovina (CAI), 2 .. plášť, 3 .. jádro

veličiny: N počet atomů M hmotnost C koncentrace $\gamma \equiv \frac{M_3}{M}$ hmotnostní poměr
jádra $D \equiv \frac{C_3}{C_2}$ rozdělovací koeficient ¹rozhraní jádro/plášť (CMB)

1.1 Geochemické rovnice

transport stabilních izotopů zákon zachování wolframu Jacobsen (2009)

$$\dot{N}_1 + \dot{N}_2 + \dot{N}_3 = 0$$

$$\dot{N}_1 = -C_1 \dot{M} \quad (1)$$

$$\dot{N}_2 = -\dot{N}_1 - \dot{N}_3 = C_1 \dot{M} - DC_2 \dot{M}_3 = (C_1 - \gamma DC_2) \dot{M} \quad (2)$$

$$\dot{N}_3 = \gamma DC_2 \dot{M} \quad (3)$$

koncentrace

$$C_2 = \frac{N_2}{M_2}$$

$$\dot{C}_2 = \dot{N}_2 \frac{1}{M_2} - N_2 \frac{1}{M_2^2} \dot{M}_2 = \frac{C_1 - (\gamma D + 1 - \gamma) C_2}{(1 - \gamma) M} \dot{M}$$

frakcionace Hf/W

$$f_2 \equiv \frac{N_{\text{Hf},2}/N_{\text{W},2}}{N_{\text{Hf},1}/N_{\text{W},1}} - 1 = \frac{N_{\text{W},1}}{N_{\text{W},2}} - 1 = \frac{N_{\text{W},1} - N_{\text{W},2}}{N_{\text{W},2}} = \frac{N_{\text{W},3}}{N_{\text{W},2}} \left(= \frac{N_3}{N_2} \right)$$

neboť Hf je silně litofilní ($N_{\text{Hf},3} \doteq 0$) stabilní izotopy ¹⁸⁰Hf, ¹⁸³W

$$f_2 = \frac{\dot{N}_3}{N_2} - N_3 \frac{1}{N_2^2} \dot{N}_2,$$

dle (2), (3):

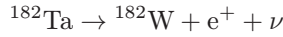
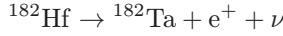
$$\dot{f}_2 = \gamma DC_2 \dot{M} \frac{1}{C_2 M_2} - \frac{N_3}{N_2} \frac{1}{C_2 M_2} (C_1 - \gamma DC_2) \dot{M} = \frac{\gamma D - f_2 \left(\frac{C_1}{C_2} - \gamma D \right)}{(1 - \gamma) M} \dot{M}$$

¹ je-li $D(t)$ a $\dot{M} > 0$, neplatí to neustále, ale jen pro $\dot{M}$

radiogenní izotop ^{182}W odchylka od mlhoviny (tzv. chondritického uniformního rezervoáru; CHUR)

$$\varepsilon_2 \equiv 10^4 \left(\frac{N_{182\text{W},2}/N_{183\text{W},2}}{N_{182\text{W},1}/N_{183\text{W},1}} - 1 \right)$$

radioaktivní rozpady (s poločasy $T_{1/2} = 9 \text{ Myr}$ a 144 d):



rozpadový zákon

$$\dot{N}_{182\text{Hf}} = -\lambda N_{182\text{Hf}}$$

$$\dot{N}_{182\text{W}} = +\lambda N_{182\text{Hf}}$$

$$N_{182\text{Hf}} = N_{182\text{Hf}}(0) e^{-\lambda t}$$

$$N_{182\text{W}} = N_{182\text{W}}(0) + N_{182\text{Hf}}(0)(1 - e^{-\lambda t}) \doteq N_{182\text{W}}(0)$$

jen několik ppm pak lze odvodit (Jacobsen 2005):

$$\dot{\varepsilon}_2 = Q^* f_2 - \frac{C_1}{C_2(1-\gamma)M} \dot{M} \varepsilon_2$$

$$Q^* = 10^4 \lambda e^{-\lambda t} \frac{C_{182\text{Hf},1}(0)}{C_{180\text{Hf},1}(0)} \frac{C_{180\text{Hf},1}(0)}{C_{182\text{W},1}(t)}$$

$C_{182\text{W},1}(t) \doteq C_{182\text{W},1}(0)$; liší se jen o několik ε

$$\dot{N}_{182\text{W},2} = +\lambda N_{182\text{Hf},2} + (C_1 - \gamma DC_2) \dot{M},$$

$$\dot{N}_{182\text{W},1} = +\lambda N_{182\text{Hf},1} - C_1 \dot{M},$$

$$\dot{N}_{183\text{W},1} = -C_1 \dot{M},$$

$$\dot{N}_{183\text{W},2} = (C_1 - \gamma DC_2) \dot{M},$$

$$\begin{aligned} \dot{\varepsilon}_2 &= 10^4 \overbrace{\frac{N_{182\text{W},2}}{N_{182\text{W},1}} \frac{N_{183\text{W},1}}{N_{183\text{W},2}}}^{\doteq 1} \left(\frac{\dot{N}_{182\text{W},2}}{N_{182\text{W},2}} - \frac{\dot{N}_{182\text{W},1}}{N_{182\text{W},1}} + \frac{\dot{N}_{183\text{W},1}}{N_{183\text{W},1}} - \frac{\dot{N}_{183\text{W},2}}{N_{183\text{W},2}} \right) = \\ &= \dots \left(\lambda \frac{N_{182\text{Hf},2}}{N_{182\text{W},2}} + \frac{C_{182\text{W},1} \dot{M}}{N_{182\text{W},2}} - \frac{\gamma DC_{182\text{W},2} \dot{M}}{N_{182\text{W},2}} - \lambda \frac{N_{182\text{W},1}}{N_{182\text{W},1}} + \frac{C_{182\text{W},1} \dot{M}}{N_{182\text{W},1}} - \right. \\ &\quad \left. - \frac{C_{183\text{W},1} \dot{M}}{N_{183\text{W},1}} - \frac{C_{183\text{W},1} \dot{M}}{N_{183\text{W},2}} + \frac{\gamma DC_{183\text{W},2} \dot{M}}{N_{183\text{W},2}} \right) = \\ &= \dots \lambda \left(\frac{N_{182\text{Hf},2}}{N_{182\text{W},2}} - \frac{N_{182\text{W},1}}{N_{182\text{W},1}} \right) + \dots \dot{M} \left(\frac{C_{182\text{W},1}}{N_{182\text{W},2}} - \frac{C_{183\text{W},1}}{N_{183\text{W},2}} \right) = \\ &= 10^4 \lambda \frac{C_{182\text{Hf},1}(0) e^{-\lambda t}}{C_{182\text{W},1}(t)} \frac{C_{180\text{Hf},1}}{C_{180\text{Hf},1}} \left(\frac{N_{182\text{Hf},2}}{N_{182\text{W},2}} \frac{N_{182\text{W},1}}{N_{182\text{Hf},1}} - 1 \right) + \\ &+ 10^4 \frac{C_{183\text{W},1}}{C_{183\text{W},2}} \frac{1}{(1-\gamma)M} \dot{M} \left(\frac{C_{182\text{W},1}}{C_{182\text{W},2}} \frac{C_{183\text{W},2}}{C_{183\text{W},1}} - 1 \right) = \\ &= Q^* f_2 + \frac{C_1}{C_2(1-\gamma)M} \dot{M} \varepsilon_2 \quad \text{c.b.d.} \end{aligned}$$

počáteční podmínky, rovnováha

$$C_2 = \frac{C_1}{\gamma D + 1 - \gamma}$$

$$f_2 = \frac{\gamma D}{1 - \gamma}$$

$$\varepsilon_2 = 0$$

1.2 Diferenciace Země

1.3 Měsíční impakt

definice

$$a \equiv \frac{1}{1 - \gamma} \frac{C_1}{C_2} = \frac{\gamma D + 1 - \gamma}{1 - \gamma} = \frac{\gamma D}{1 - \gamma} + 1 = f_2 + 1$$

za prvé vezměme ε_2 , bez radioaktivity

$$\dot{\varepsilon}_2 \doteq -a \frac{\dot{M}}{M} \varepsilon_2$$

$$\frac{\dot{\varepsilon}_2}{\varepsilon_2} = -a \frac{\dot{M}}{M}$$

neurčitý integrál

$$\ln \varepsilon_2 = -a \ln M + C$$

$$\varepsilon_2 = C' M^{-a}$$

konstanta před impaktem $C' = \varepsilon_2 M^a$ čili po impaktu

$$\varepsilon'_2 = \varepsilon_2 \left(\frac{M}{M'} \right)^a$$

obdobně

$$\dot{C}_2 = \left(\frac{C_1}{1 - \gamma} - a C_2 \right) \frac{\dot{M}}{M} \quad (4)$$

partikulární řešení

$$\dot{C}_2 = -a C_2 \frac{\dot{M}}{M}$$

totéž jako ε

$$C_2 = C' M^{-a}$$

variace konstant

$$C_2 = C'(M) M^{-a}$$

$$\frac{dC_2}{dM} = \frac{dC'}{dM} M^{-a} + C'(-a) M^{-a-1} = \left(\frac{C_1}{1-\gamma} - aC' M^{-1} \right) M^{-1}$$

kam jsme dosadili (4)

$$\frac{dC'}{dM} = \frac{C_1}{1-\gamma} M^{a-1}$$

$$C'(M) = \frac{C_1}{1-\gamma} \frac{M^a}{a} + C''$$

kde C'' je druhá integrační konstanta řešení

$$C_2 = \left(\frac{C_1}{1-\gamma} \frac{M^a}{a} + C'' \right) M^{-a}$$

má-li pro M , vyjít právě C_2 , pak:

$$C'' = C_2 M^a - \frac{C_1}{1-\gamma} \frac{M^a}{a}$$

$$C'_2 = \left(\frac{C_1}{1-\gamma} \frac{M'^a}{a} + C_2 M^a - \frac{C_1}{1-\gamma} \frac{M^a}{a} \right) M'^{-a}$$

$$C'_2 = \frac{C_1}{1-\gamma} \frac{1}{a} \left[1 - \left(\frac{M}{M'} \right)^a \right] + C_2 \left(\frac{M}{M'} \right)^a$$

obdobně obdobně

$$\dot{f}_2 = \frac{\gamma D - f_2 \left(\frac{C_1}{C_2} - \gamma D \right)}{(1-\gamma)M} \dot{M}$$

partikulárně

$$\frac{\dot{f}_2}{f_2} = -b \frac{\dot{M}}{M}$$

kde $b = (\frac{C_1}{C_2} - \gamma D)/(1-\gamma)$

$$f_2 = C' M^{-b}$$

variabilně

$$f_2 = C'(M)M^{-b}$$

$$\frac{df_2}{dM} = \frac{dC'}{dM}M^{-b} - C'bM^{-b-1} = -f_2bM^{-1} + \frac{\gamma D}{1-\gamma}M^{-1}$$

$$\frac{dC'}{dM} = \frac{\gamma D}{1-\gamma}M^{b-1}$$

$$C' = \frac{\gamma D}{1-\gamma} \frac{M^b}{b}$$

$$f_2 = \frac{\gamma D}{1-\gamma} \frac{M^b}{b} M^{-b} = \frac{\gamma D}{1-\gamma} \frac{1-\gamma}{\frac{C_1}{C_2} - \gamma D} = \frac{\gamma D}{\frac{C_1}{C_2} - \gamma D} = f_2$$

což nezávisí na M , čili $f'_2 = f_2$

1.4 Variabilní rozdělování

rozdělovací koeficient $D \neq \text{konst.}$

fugacita f_{O_2}

1.5 Akrece dle N-částicových modelů

1.6 Systém U/Pb

1.7 Systém D/H

[1] JACOBSEN, 2005.

[2] RUDGE, 2010.

[3] YU, JACOBSEN, 2011.